organic papers

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Key indicators

Single-crystal X-ray study T = 293 K Mean σ (C–C) = 0.004 Å R factor = 0.033 wR factor = 0.099 Data-to-parameter ratio = 8.9

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e. In the title compound, $C_{19}H_{29}ClO_3$, the hydroxyl linkage in ring A is equatorial. The six-membered rings A, B and C have chair conformations. The five-membered ring D adopts a 14α envelope conformation. The A/B, B/C and C/D ring junctions are *trans*. The packing of the molecules is assumed to be dictated mainly by two strong intermolecular hydrogen bonds of the type O-H···O.

Comment

Topical corticosteroids became available for medical use in the 1950's when hydrocortisone was introduced demonstrating substantial topical anti-inflamatory potency. Modification of the basic corticosteroid molecule, e.g. by halogenation or the optimization of the vehicle, enhances their effectiveness, the former by raising the potency and the latter by increasing the percutaneous absorption (Lutsky et al., 1979). The strategy and importance for the synthesis of these compounds have antecedents in similar structures, with anabolic and/or and rogenic activity, replacing positions 9α and 11β with fluorine and hydroxyl, respectively (Shapiro et al., 1987). An example of this is 9α -fluoro- 11β , 17α -dihydroxy- 17α -methyl-4androsten-3-one (halotestin), a commercial compound 20 times more androgenic and 10 times more anabolic than methyltestosterone. In connection with our studies on the synthesis and characterization of bioactive steroids, the structure of the title compound, (I), could allow the prediction of the possibility of a compound of presenting/displaying anabolic and/or androgenic properties. The absolute configuration turned out to be the same as that predicted beforehand from the synthesis route.



Fig. 1 shows the molecular structure of the title compound, with the corresponding numbering scheme. The C3–O3 bond of the hydroxy group is equatorially oriented and (–)antiperiplanar to the C4–C5 bond. The presence of OH bonded to C3 does not disturb the chair conformation in the ring A of the steroidal nucleus. Ring A has a chair conformation with all asymmetry parameters (Duax *et al.*, 1976) below 9.2 (3)°. The average magnitude of the torsion angles is

© 2001 International Union of Crystallography Printed in Great Britain – all rights reserved 9α -Chloro- 3β , 11β -dihydroxy- 5α -androstan-17-one

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54.27 (12)°. Ring *B* displays a chair conformation, as does ring *C* (Pfeiffer *et al.*, 1985). The five-membered ring *D* adopts a 14 α -envelope conformation (Altona *et al.*, 1968). The *A*/*B*, *B*/*C* and *C*/*D* ring junctions are *trans*. Bond distances and valence angles are close to expected values (Honda *et al.*, 1996). The packing of the molecules is assumed to be dictated mainly by two intermolecular hydrogen bonds (Table 1).

Experimental

The synthesis of the title compound has been described by Ruíz (1997). Crystals (m.p. 399 K) were grown by slow evaporation from ethanol.

 $D_x = 1.297 \text{ Mg m}^{-3}$

Cell parameters from 40

Cu Ka radiation

reflections

 $\mu = 2.04 \text{ mm}^{-1}$

T = 293 (2) K

 $R_{\rm int} = 0.051$ $\theta_{\rm max} = 69.2^{\circ}$

 $h = -10 \rightarrow 1$

 $k = -1 \rightarrow 10$

 $l = -15 \rightarrow 15$

3 standard reflections

every 100 reflections

intensity decay: none

Prism, colourless

 $0.64 \times 0.46 \times 0.28 \text{ mm}$

 $\theta = 10.5 - 28.0^{\circ}$

Crystal data

 $C_{19}H_{29}CIO_3$ $M_r = 340.87$ Monoclinic, P_{2_1} a = 8.3450 (3) Å b = 8.5536 (3) Å c = 12.6875 (5) Å $\beta = 105.482$ (3)° V = 872.77 (6) Å³ Z = 2

Data collection

Siemens P4 four-circle diffractometer $\omega/2\theta$ scans Absorption correction: ψ scan (North *et al.*, 1968) $T_{min} = 0.247, T_{max} = 0.356$ 2286 measured reflections 1893 independent reflections 1862 reflections with $I > 2\sigma(I)$

Refinement

Refinement on F^2 $w = 1/[\sigma^2(F_o^2) + (0.0628P)^2]$ $R[F^2 > 2\sigma(F^2)] = 0.033$ + 0.1216P] $wR(F^2) = 0.099$ where $P = (F_o^2 + 2F_c^2)/3$ S = 1.10 $(\Delta/\sigma)_{\rm max} = 0.009$ $\Delta \rho_{\rm max} = 0.21 \ {\rm e} \ {\rm \AA}^{-3}$ 1893 reflections $\Delta \rho_{\rm min} = -0.28 \text{ e } \text{\AA}^{-3}$ 213 parameters H atoms treated by a mixture of Extinction correction: SHELXL97 independent and constrained Extinction coefficient: 0.032 (2) Absolute structure: Flack (1983) refinement Flack parameter = 0.012(17)

Table 1

Hydrogen-bonding geometry (Å, °).

$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdot \cdot \cdot A$	$D - \mathbf{H} \cdot \cdot \cdot A$
O3−H3A···O17 ⁱ	0.82	2.08	2.874 (3)	164
$O11-H11A\cdots O3^{ii}$	0.82	2.05	2.862 (3)	171
8 (1)	1 1	1 1		

Symmetry codes: (i) x - 1, 1 + y, z; (ii) $1 - x, y - \frac{1}{2}, -z$.

H atoms were calculated geometrically and included in the refinement, but were constrained to ride on their parent atoms. The isotropic displacement parameters of the H atoms were fixed to $1.3U_{eq}$ of their parent atoms. The torsion angles about the C–OH and C–CH₃ bonds were allowed to refine. Number of Friedel pairs used: 232. Friedel pairs were treated as independent data.





Plot showing the atomic numbering scheme for the title compound. Displacement ellipsoids are drawn at 50% probability level for non-H atoms.

Data collection: *XSCANS* (Siemens,1996); cell refinement: *XSCANS*; data reduction: *XSCANS*; program(s) used to solve structure: *SIR*92 (Altomare *et al.*, 1994); program(s) used to refine structure: *SHELXL*97 (Sheldrick, 1997); molecular graphics: *DIAMOND* (Bergerhoff, 1996); software used to prepare material for publication: *PLATON* (Spek, 1990), *PARST* (Nardelli, 1983, 1995) and *PARSTCIF* (Nardelli, 1991).

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