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## Structure Reports

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## Key indicators

Single-crystal X-ray study
$T=293 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.004 \AA$
$R$ factor $=0.033$
$w R$ factor $=0.099$
Data-to-parameter ratio $=8.9$

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.
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# $9 \alpha$-Chloro-3 $\beta, 11 \beta$-dihydroxy- $5 \alpha$-androstan-17-one 

In the title compound, $\mathrm{C}_{19} \mathrm{H}_{29} \mathrm{ClO}_{3}$, the hydroxyl linkage in ring $A$ is equatorial. The six-membered rings $A, B$ and $C$ have chair conformations. The five-membered ring $D$ adopts a $14 \alpha$ envelope conformation. The $A / B, B / C$ and $C / D$ ring junctions are trans. The packing of the molecules is assumed to be dictated mainly by two strong intermolecular hydrogen bonds of the type $\mathrm{O}-\mathrm{H} \cdots \mathrm{O}$.

## Comment

Topical corticosteroids became available for medical use in the 1950's when hydrocortisone was introduced demonstrating substantial topical anti-inflamatory potency. Modification of the basic corticosteroid molecule, e.g. by halogenation or the optimization of the vehicle, enhances their effectiveness, the former by raising the potency and the latter by increasing the percutaneous absorption (Lutsky et al., 1979). The strategy and importance for the synthesis of these compounds have antecedents in similar structures, with anabolic and/or androgenic activity, replacing positions $9 \alpha$ and $11 \beta$ with fluorine and hydroxyl, respectively (Shapiro et al., 1987). An example of this is $9 \alpha$-fluoro- $11 \beta, 17 \alpha$-dihydroxy- $17 \alpha$-methyl-4-androsten-3-one (halotestin), a commercial compound 20 times more androgenic and 10 times more anabolic than methyltestosterone. In connection with our studies on the synthesis and characterization of bioactive steroids, the structure of the title compound, (I), could allow the prediction of the possibility of a compound of presenting/displaying anabolic and/or androgenic properties. The absolute configuration turned out to be the same as that predicted beforehand from the synthesis route.

(I)

Fig. 1 shows the molecular structure of the title compound, with the corresponding numbering scheme. The $\mathrm{C} 3-\mathrm{O} 3$ bond of the hydroxy group is equatorially oriented and ( - )antiperiplanar to the $\mathrm{C} 4-\mathrm{C} 5$ bond. The presence of OH bonded to C3 does not disturb the chair conformation in the ring $A$ of the steroidal nucleus. Ring $A$ has a chair conformation with all asymmetry parameters (Duax et al., 1976) below $9.2(3)^{\circ}$. The average magnitude of the torsion angles is

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54.27 (12) ${ }^{\circ}$. Ring $B$ displays a chair conformation, as does ring $C$ (Pfeiffer et al., 1985). The five-membered ring $D$ adopts a $14 \alpha$-envelope conformation (Altona et al., 1968). The $A / B, B /$ $C$ and $C / D$ ring junctions are trans. Bond distances and valence angles are close to expected values (Honda et al., 1996). The packing of the molecules is assumed to be dictated mainly by two intermolecular hydrogen bonds (Table 1).

## Experimental

The synthesis of the title compound has been described by Ruíz (1997). Crystals (m.p. 399 K ) were grown by slow evaporation from ethanol.

## Crystal data

$\mathrm{C}_{19} \mathrm{H}_{29} \mathrm{ClO}_{3}$
$M_{r}=340.87$
Monoclinic, $P 2_{1}$
$a=8.3450(3) \AA$
$b=8.5536(3) \AA$
$c=12.6875(5) \AA$
$\beta=10.482(3){ }^{\circ}$
$V=872.77(6) \AA^{3}$
$Z=2$

$$
\begin{aligned}
& D_{x}=1.297 \mathrm{Mg} \mathrm{~m}^{-3} \\
& \text { Cu Ka radiation } \\
& \text { Cell parameters from } 40 \\
& \text { reflections } \\
& \theta=10.5-28.0^{\circ} \\
& \mu=2.04 \mathrm{~mm}^{-1} \\
& T=23(2) \mathrm{K} \\
& \text { Prism, colourless } \\
& 0.64 \times 0.46 \times 0.28 \mathrm{~mm}
\end{aligned}
$$

## Data collection

Siemens $P 4$ four-circle diffractometer
$\omega / 2 \theta$ scans
Absorption correction: $\psi$ scan
(North et al., 1968)
$T_{\text {min }}=0.247, T_{\text {max }}=0.356$
2286 measured reflections
1893 independent reflections
1862 reflections with $I>2 \sigma(I)$

## Refinement

Refinement on $F^{2}$
$R_{\text {int }}=0.051$
$\theta_{\text {max }}=69.2^{\circ}$
$h=-10 \rightarrow 1$
$k=-1 \rightarrow 10$
$l=-15 \rightarrow 15$
3 standard reflections every 100 reflections intensity decay: none
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.033$
$w R\left(F^{2}\right)=0.099$
$S=1.10$
1893 reflections
213 parameters
H atoms treated by a mixture of independent and constrained refinement
$w=1 /\left[\sigma^{2}\left(F_{o}{ }^{2}\right)+(0.0628 P)^{2}\right.$ $+0.1216 P$ ]
where $P=\left(F_{o}{ }^{2}+2 F_{c}^{2}\right) / 3$
$(\Delta / \sigma)_{\max }=0.009$
$\Delta \rho_{\max }=0.21 \mathrm{e}_{\AA^{-3}}$
$\Delta \rho_{\min }=-0.28 \mathrm{e}^{\AA^{-3}}$
Extinction correction: SHELXL97
Extinction coefficient: 0.032 (2)
Absolute structure: Flack (1983)
Flack parameter $=0.012(17)$

Table 1
Hydrogen-bonding geometry $\left(\AA,{ }^{\circ}\right)$.

| $D-\mathrm{H} \cdots A$ | $D-\mathrm{H}$ | $\mathrm{H} \cdots A$ | $D \cdots A$ | $D-\mathrm{H} \cdots A$ |
| :--- | :--- | :--- | :--- | :--- |
| ${\text { O3-H3A } \cdots \mathrm{O}^{2} 7^{\mathrm{i}}}^{\mathrm{ii}}$ | 0.82 | 2.08 | $2.874(3)$ | 164 |
| ${\text { O11-H11 } A \cdots \mathrm{O}^{2}}^{0.82}$ | 0.05 | $2.862(3)$ | 171 |  |

Symmetry codes: (i) $x-1,1+y, z$; (ii) $1-x, y-\frac{1}{2},-z$.
H atoms were calculated geometrically and included in the refinement, but were constrained to ride on their parent atoms. The isotropic displacement parameters of the H atoms were fixed to $1.3 U_{\text {eq }}$ of their parent atoms. The torsion angles about the $\mathrm{C}-\mathrm{OH}$ and $\mathrm{C}-\mathrm{CH}_{3}$ bonds were allowed to refine. Number of Friedel pairs used: 232. Friedel pairs were treated as independent data.


Figure 1
Plot showing the atomic numbering scheme for the title compound. Displacement ellipsoids are drawn at $50 \%$ probability level for non-H atoms.

Data collection: XSCANS (Siemens,1996); cell refinement: $X S C A N S$; data reduction: XSCANS; program(s) used to solve structure: SIR92 (Altomare et al., 1994); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: DIAMOND (Bergerhoff, 1996); software used to prepare material for publication: PLATON (Spek, 1990), PARST (Nardelli, 1983, 1995) and PARSTCIF (Nardelli, 1991).

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## References

Altomare, A., Cascarano, G., Giacovazzo, C., Guagliardi, A., Burla, M. C., Polidori, G. \& Camalli, M. (1994). J. Appl. Cryst. 27, 435.
Altona, C., Geize, H. J. \& Romers, C. (1968). Tetrahedron, 24, 13-32.
Bergerhoff, G. (1996). DIAMOND. Gerhard-Domagk-Straße 1, 53121 Bonn, Germany.
Duax, W. L., Weeks, C. M. \& Roher, D. C. (1976). Topics in Stereochemistry, Vol. 9, edited by E. L. Eliel \& N. Allinger, pp. 271-383. New York: John Wiley.
Flack, H. D. (1983). Acta Cryst. A39, 876-881.
Honda, T., Fujii, I., Hirayama, N., Ishikawa, D., Kawagishi, H., Song, K.-S. \& Yoo, I.-D. (1996). Acta Cryst. C52, 1550-1552.
Lutsky, B., Berkenkopf, J., Fernandez, X., Monahan, M., Shue, H. J., Tiberi, R. L. \& Green, M. J. (1979). Arzneim. Forsch. 29, 1662-1667.

Nardelli, M. (1983). Comput. Chem. 7, 95-98.
Nardelli, M. (1991). PARSTCIF. University of Parma, Italy.
Nardelli, M. (1995). J. Appl. Cryst. 28, 659.
North, A. C. T., Phillips, D. C. \& Mathews, F. S. (1968). Acta Cryst. A24, 351359.

Pfeiffer, D., Kutschabsky, L., Kretschmer, R. G., Coll, F. \& Adam, G. (1985). Z. Chem. 25, 183-184.
Ruíz, J. A. (1997). Sintesis del acetato de 5 $\alpha$-9(11)-androsten-3 $\beta$-ol-17-ona a partir de Dieno. Rev. Cub. Farm. Fondo Nacional de Manuscritos Cientificos del Inst. de Docum. Científico Técnico de la Academia de Ciencias de Cuba. (In Spanish.)
Shapiro, E. L., Gentles, M. J., Tiberi, R. L. Popper, T. L., Berkenkopf J., Lutsky, B. \& Watnick, A. S. (1987). J. Med. Chem. 30, 1068-1073.

Sheldrick, G. M. (1997). SHELXL97. University of Göttingen, Germany.
Siemens (1996). XSCANS. Version 2.2. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
Spek, A. L. (1990). Acta Cryst. A46, C-34.

